



PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : A61K 7/42	A1	(11) International Publication Number: WO 96/01617 (43) International Publication Date: 25 January 1996 (25.01.96)
(21) International Application Number: PCT/US95/08572 (22) International Filing Date: 10 July 1995 (10.07.95) (30) Priority Data: 08/272,096 8 July 1994 (08.07.94) US (60) Parent Application or Grant (63) Related by Continuation US 08/272,096 (CIP) Filed on 8 July 1994 (08.07.94) (71) Applicant (for all designated States except US): APOGEN [US/US]; 8704 Melrose Lane, River Ridge, LA 70123 (US). (71)(72) Applicants and Inventors: LI, Yin-Xiong [CN/US]; Apartment H3, 1011 Hickman Road, Augusta, GA 30904 (US). SHI, Ergang [CN/US]; Apartment B-10, 406 Aboca Street, Nashville, TN 37203 (US). (74) Agent: BRUESS, Steven, C.; Merchant, Gould, Smith, Edell, Welter & Schmidt, P.A., 3100 Norwest Center, 90 South Seventh Street, Minneapolis, MN 55402-4131 (US).		(81) Designated States: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LT, LU, LV, MD, MG, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TT, UA, UG, US, UZ, VN, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG), ARIPO patent (KE, MW, SD, SZ, UG). Published <i>With international search report.</i>
(54) Title: NUCLEIC ACID FILTERS (57) Abstract The use of nucleic acid-containing materials, such as deoxyribonucleic acid and ribonucleic acid, as ultraviolet radiation filters for specifically absorbing genetic hazard ultraviolet radiation by the interposition of a nucleic acid-containing barrier between a source of UV radiation and a living organism.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgyzstan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovakia
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechoslovakia	LV	Larvia	TG	Togo
CZ	Czech Republic	MC	Monaco	TJ	Tajikistan
DE	Germany	MD	Republic of Moldova	TT	Trinidad and Tobago
DK	Denmark	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	US	United States of America
FI	Finland	MN	Mongolia	UZ	Uzbekistan
FR	France			VN	Viet Nam
GA	Gabon				

5

TITLE: NUCLEIC ACID FILTERS

SPECIFICATION:

10

BACKGROUND OF THE INVENTION:

1. Field of the Invention:

The present invention relates to the use of nucleic acids such as deoxyribonucleic acid and
15 ribonucleic acid (hereinafter collectively referred to as "NA") as a filter for removing environmental hazards that otherwise would pose a threat to the genetic material of living organisms such as plants and animals, including humans. More particularly, the
20 present invention relates to a method of use of NA as a sunlight or artificial UV light filter to selectively remove nucleic acid-damaging ultraviolet radiation, for example, from sunlight or artificial UV source, before it contacts a living organism. Even more particularly,
25 the present invention relates to NA-containing compositions and articles which form an ultraviolet radiation-absorbing layer between the sun and living organisms where the layer specifically blocks nucleic acid-damaging ultraviolet radiation before it contacts
30 a living organism.

2. Background of the Invention:

One of the major causes of cancer is the alteration of the cellular genome. Environmental hazards that specifically react with nucleic acids are
35 a major cause of these genetic alterations, and hence a major cause of cancer in both plants and animals. Typical environmental hazards that have nucleic acid-damaging capabilities include, inter alia, ultraviolet radiation, ionizing radiation, and environmental

chemicals that display either nucleic acid intercalating activity or the ability to form adducts with nucleic acids.

Exposure to ultraviolet radiation from the sun is generally believed to be the major cause of non-melanoma skin cancers and a major cause of malignant melanoma skin cancers (Moan and Dahlback, Br. J. Cancer 65:916-921, 1992). Rates of both non-melanoma and malignant melanoma skin cancers have increased dramatically in recent years. For example, during the period between 1957 and 1984 in Norway, the incidence of cutaneous malignant melanoma increased by 350% for men and 440% for women. According to a recent U.S. federal survey taken between 1977-1988, about 500,000 basal-cell carcinomas and 100,000 squamous-cell carcinomas occur annually. Other, smaller surveys suggest the incidence may have increased by as much as 65% since 1980. (Preston and Stern, New England J. Med., 327:1649-1662, 1992). Scientists have linked the recent increase in skin cancer with the decrease in the Earth's protective ozone layer, a layer that normally filters out much of the nucleotide damaging ultraviolet radiation in sunlight. As the Earth's protective ozone layer continues to decrease, the earth's surface is exposed to increasingly higher levels of nucleic acid-damaging ultraviolet radiation from the sun. Ozone depletion may seriously impact such important biological end-points as skin cancer, cataracts, the immune system, crop yields, and oceanic phytoplankton (Coohill, Photochemistry and Photobiol., 54:859-870, 1991). Scientists believe that this increased irradiation will especially lead to a related increase the incidence of sunlight-induced skin cancer. (Kelfkens G., etc. Photochem. Photobiol., 52(4):819-23, 1990). Henriksen predicted a two percent increase in

overall skin cancer for each one percent depletion of stratospheric ozone. (Henriksen T., Photochem. Photobiol. 51(5):579-82, 1990).

Currently, in response to the threat of exposure
5 to harmful ultraviolet radiation from the sun, people are advised to avoid excessive exposure to sunlight, and when in the sun, to wear sunscreens. The shortcomings of these actions are many fold, including their limitation to humans, as opposed to all other
10 living creatures, as well as numerous problems inherent in existing sunscreen technology.

A more desired response to the global threat of increased exposure to nucleic acid-damaging ultraviolet radiation would provide protection for all living
15 creatures, not merely humans. The present invention affords such protection.

Existing topical sunscreens fall within one of two categories: (1) chemical, and (2) physical sunscreens. Physical sunscreens operate by blocking out nearly all
20 wavelengths of sunlight (290-760 nm). They are typically applied in an opaque cream or lotion containing particulate ingredients that do not selectively absorb ultraviolet radiation, but when applied as a thin film, primarily reflect and scatter
25 sunlight. Common ingredients in physical sunscreens include titanium dioxide, zinc oxide and talc.

Because they completely block all sunlight, known physical sunscreens prevent the beneficial effects of sunlight upon the skin. For example, physical
30 sunscreens block the wavelengths of sunlight that are essential for the skin to convert 7-dehydro-cholesterol into vitamin D₃ (Matsuoka et.al., Arch. Dermatol. 124:1802-1804, 1988) and block sunlight required for the photoactivation of a protein called photolyase
35 which is involved in cellular DNA repair mechanism

(Rupert, C.S., J.Gen.Physiol. 45:725-741, 1962).

Physical sunscreens also prevent the often desired "tanning" of the skin which would otherwise occur upon exposure to moderate amounts of sunlight. Further, physical sunscreen formulations are cosmetically unpleasing, unacceptable to many patients, and messy to use.

Chemical sunscreens contain one or more ultraviolet-absorbing chemicals which are generally colorless, and upon application of a thin and invisible film, act as filters to selectively prevent certain wavelengths of ultraviolet light from reaching the cells of the epidermis. However, since chemical sunscreens are non-natural molecules, their molecular structures are quite different from the nucleic acid molecules which are the direct target for ultraviolet light-induced damage to living organisms (Ananthaswamy and Pierceall, Photochem and Photobiol., 52(6):1118-1136, 1990). Also, due to a lack of understanding of the photochemistry of nucleic acids, whether or not chemical sunscreens protect living organisms from skin cancer is still an unresolved controversy (Gurish, J. Invest. Dermatol., 76:246-251, 1981; Knowland, FEBS Letters, 324:308-313, 1993; Thompson, The New England J. of Med., 328:1147-1151, 1993). Further, the radiation that is blocked out by known chemical sunscreens often includes frequencies of sunlight that are beneficial to the body such as the wavelengths of ultraviolet light that are necessary for the body to produce vitamin D.

Most commercial sunscreens contain at least one each of two different types of chemicals: UV-A-absorbing chemicals and UV-B-absorbing chemicals. UV-A-absorbing chemicals absorb ultraviolet radiation in the range of 320-400nm. UV-B-absorbing chemicals

absorb ultraviolet radiation in the range of 290-320nm. The wavelength range of UV-A and UV-B (290-400nm) includes the frequencies of ultraviolet light that cause sunburn (290-350nm), however, it also includes beneficial wavelengths of ultraviolet light such as those that are necessary for the skin's production of vitamin D₃ (290-315nm) and also those required for repair of pyrimidine dimers (320-600 nm) (Spikes, J.D., In: Experimental and Clinical Photoimmunology, Daynes and Spikes (eds.), CRC Press, Boca Raton (1983); Chiang and Rupert, Photochem. Photobiol., 30:525-528, 1979; Hanawalt and Haynes, Scientific American, 216:36, 1967).

UV-absorbing chemicals include para-aminobenzoic acid (PABA) and esters thereof, benzophenones, and cinnemates which selectively absorb and screen sunburn-producing UV radiation (290-320 nm). However, sunscreens containing these chemicals can cause selective burning (smarting), contact dermatitis, and allergic contact dermatitis. (Kaidbey et al., J. Soc. Cosmet. Chem. 29:525-536, 1978; Fisher, AA, Arch. Dermatol., 113: 1288-1300, 1977; Toby-Mathias et al., Arch Dermatol., 114:1665-1666, 1978). Further, synthetic procedures for producing UV-absorbing chemicals may introduce contaminants into the preparation, some which may be carcinogenic. For example, urocranic acid was commercially available as a sunscreen to block UV-B, but was removed from the market because of its link to skin cancer (Consumer Reports, June 1991, p. 406). Chemical sunscreens may mutate on exposure to sunlight, the mutagen being carcinogenic. Indeed, studies suggest that chemical sunscreens might encourage rather than prevent sunlight-related cancers (Knowland et al., FEBS Letters, 324:309-313, 1993).

Further, the sun protection factor (SPF) rating used to quantitate protection offered by a commercial sunscreen product is based upon erythema, a morphological criteria that is not directly related to quantitation of genetic hazard protection. The SPF rating may be more harmful than useful. Those who purchase and use a sunscreen with a high SPF expect the sunscreen to protect them from genetic hazard UV radiation. With a false sense of security, they may spend a greater amount of time exposed to UV radiation, while the sunscreen may not provide adequate protection. (Young, British J. of Dermatology, 122, Supplement 35:111-114, 1990; Young, Pigment Cell Research, 1:350-4, 1988). The SPF rating is not directly related in any way to UV-induced DNA damage. The SPF standard is based upon errythmia, or redness caused by UV exposure.

The wavelengths of light absorbed by the various chemical sunscreens differs with the compounds used. Because the specific wavelengths responsible for UV-induced DNA damage are not fully understood, a non-target compound cannot be certain to be an effective filter.

It would be of great utility to provide a natural genetic hazard UV radiation filter which could selectively and specifically eliminate genetic-hazard inducing UV radiation.

SUMMARY OF THE INVENTION:

The present invention describes a method wherein nucleic acids are used to selectively remove nucleic acid-damaging hazards from the environment before those hazards would otherwise reach and potentially harm plants and animals, including humans.

In the preferred embodiment of the invention, NA is used to selectively absorb, and thereby remove, nucleic acid-damaging ultraviolet radiation from artificial sources or from sunlight. Generally, in the preferred embodiment, a layer of NA-containing material is interposed between a living organism and the UV source the NA-containing material may be, for example, a cream applied to the surface of a target; a fabric sheet worn as clothing by the target, or a shield protecting a larger habitat. Generally, this NA-containing barrier contains a sufficient amount of nucleic acid to selectively absorb and filter genetic hazard ultraviolet radiation and thereby shield a target from nucleic acid-damaging radiation. This NA-containing barrier generally consists of, for example, NA in or on a transparent surface, in or on a solid matrix, a NA coated fiber which may or may not be woven, a composition for topical application of an NA-containing mixture, or other suitable means of placing an NA-containing barrier between a UV radiation source (sun or artificial UV source) and a living organism. In a preferred embodiment, a NA-containing lotion or cream is applied to a surface such as skin to shield an animal from exposure to damaging ultraviolet radiation.

The preferred embodiment of the present invention overcomes problems presented by physical sunscreens and by chemical sunscreens. Because the nucleic acid filter reacts with ultraviolet radiation in approximately the same way as UV contacts and catalyzes DNA molecules in living organisms, its use as a sunlight filter is clearly superior to existing chemical and physical sunscreens. By using the same target molecules in the filter, the nucleic acid filter provides selective absorption of the specific UV spectra that cause cancer-inducing genetic damage,

while allowing other, beneficial wavelengths of sunlight to shine upon epidermal cells. In an alternative preferred embodiment, NA is applied in or on the surface of glass barrier, such as eyeglasses, to protect the
5 wearer's eyes from damaging effects of the sun.

DETAILED DESCRIPTION:

The present invention relates to a method of filtering out or absorbing genetic hazard ultraviolet radiation before it contacts a living organism,
10 particularly nucleic acid-damaging radiation from natural or artificial UV sources, through the use of a nucleic acid-containing filter located between the UV source and a living organism. In the present invention,
15 the term "nucleic acid" (NA) is meant to include DNA and RNA as well as their constituent nucleotides, nucleosides, bases and derivatives thereof, to the extent each of these is able to selectively absorb genetic hazard UV radiation. Derivatives include, for
20 example, alcohol-modified NA due to a shift in its absorbance spectrum toward known genetic hazard wavelengths, may be preferred. The NA may be natural or synthetic, including NA produced by genetic engineering methods. One of skill in the art will recognize many
25 types of NA molecules useful as NA filters in the present invention. To screen for a useful NA molecule, one can prepare a composition of the NA and test its ability to shield a test sample of DNA from UV-induced damage, as compared with a non-shielded control, for
30 example using the tests described in the Examples which follow.

The term "genetic hazard" ultraviolet radiation is meant to include those wavelengths of ultraviolet radiation that induce alterations in genetic materials,
35 particularly in NA, including dimerization of bases.

In a preferred embodiment, the nucleic acids in the NA-barrier are either deoxyribonucleic acid ("DNA") or ribonucleic acid ("RNA"), obtained from animal and

plant tissues or by genetic engineering methods. The NA may be present in the NA barrier in any means suitable for providing a layer of NA between the UV source and a living organism. The barrier may be a
5 coating composition suitable for coating the surface of an inert substance with an even layer of NA sufficient to absorb genetic hazard ultraviolet radiation. This barrier film of NA may be in a cream for topical application to the skin, in or on an inert surface or
10 shield, in or on a solid matrix, or in or on a fiber which may or may not be woven.

The amount of NA needed to provide the desired protection will vary with the nature of the barrier, the source and intensity of the radiation and other
15 well recognized variables. Suitable amounts can be readily determined by standard methods of testing, for example, by the methods outlined in the Examples. NA is generally incorporated in the barrier in an amount ranging from about 15 $\mu\text{g}/\text{cm}^2$ to about 150 $\mu\text{g}/\text{cm}^2$,
20 preferably at least about 50 $\mu\text{g}/\text{cm}^2$.

For surface application, carriers include any vehicle or medium capable of incorporating the NA ultraviolet filter compound in a manner permitting uniform application to the surface. For topical
25 application on skin, the carrier must be pharmaceutically acceptable. The term "pharmaceutically acceptable" is intended to mean the carrier is dermatologically acceptable. The carrier may comprise a wax, oil or cream base material in which
30 NA can be held in a clear solution or a uniform dispersion, for example, as submicron size particles. Preferably the carrier comprises a suitable solvent or a mixture of solvents capable of solubilizing the NA to provide a concentration that is effective as a UV light
35 filtering agent when incorporated in the formulation.

Solvents that may be useful include water, alcohols, ketones, esters, polyesters, such as oils, hydrocarbons, ethers, polyethers, polyetherpolyols, and other special solvents such as dimethylsulfoxide, dimethylformamide, dimethylisosorbide, and the like. Some of the above-named ingredients are not pharmaceutically acceptable, but are useful in other applications.

The topical compositions may be applied as a liquid or as a lotion or cream comprising a water-in-oil, an oil-in-water or a multiple emulsion. The topical compositions will form a continuous film of genetic hazard ultraviolet radiation-filtering NA. Such film may also contain other ultraviolet absorbing chemicals, such as UV-A-absorbing and UV-B-absorbing chemicals to provide protection against sunburn. Because the UV filters are often used in hot weather and in conjunction with water-related activities, it is desirable that the protective coating applied to the skin not be appreciably affected by water or perspiration.

In addition to its application to surfaces in the form of lotions and creams, NA may also be directly applied to inert surfaces. For example, the NA may be directly annealed to plastic and to solid matrices as described more fully in the examples below.

The UV-absorbing compositions and articles of the present invention are useful as specific genetic hazard UV radiation-absorbing barriers or shields when placed between a target and a source of UV radiation. The precise location of the barrier or shield will vary, e.g., creams applied directly to the skin and insert substances as habitat shields positioned a large distance from the target. However, the placement of the barrier or shield is in the path from UV source to

target, thus providing a genetic-hazard UV radiation absorbing barrier preventing the UV radiation from contacting the target. In a preferred embodiment, the barrier is NA in or on protective eyewear such as glasses, in order to protect the target, the wearer's eyes, from damaging effects of sunlight or UV radiation.

Because the compositions of the present invention include the molecules which are a natural target for genetic-hazard UV radiation, they form a specific barrier which will absorb genetic-hazard UV radiation at whatever wavelengths, known or unknown, which cause damage. These compositions thus provide a uniquely specific barrier.

The following examples serve as non-limiting illustrations of the types of compounds and methods used in the invention. The following examples serve to illustrate, but not to limit the practice of the invention.

EXAMPLES

EXAMPLE 1

Representative Procedures for Preparation of Nucleic Acid from Animal Tissue

A. Preparation of Genomic DNA-Method I

Five volumes of homogenizing solution consisting of 10 mM Tris[Hydroxymethyl] aminomethane hydrochloride (Tris.Cl) pH 7.4, 10 mM NaCl, 25 mM ethylene diamine-tetraacetic acid (EDTA), and 0.5% sodium dodecyl sulfate (SDS), was added to one volume of fresh porcine liver which had been cut into small pieces. The resulting mixture was homogenized to break the cell membranes. The homogenate was centrifuged for 5 min at 2500 x relative centrifugal force (RCF). Then, 3 volumes of an extraction buffer consisting of 10 mM Tris.Cl pH 8.0, 0.1 M EDTA, pH 8.0, 20 ug/ml pancreatic RNase, and 0.5% SDS, were added to the supernatant. Proteinase K was then added to a final concentration of 100 ug/ml, by gently mixing the enzyme into the solution with a glass rod. The lysate was

then incubated in a water bath for 3 hours at 50°C with occasional stirring. Following the 3 hour incubation, an equal volume of phenol, equilibrated with 0.5M Tris Cl, pH8.0, was added to the solution. The two phases
5 were mixed by stirring for 10 minutes. The phases were separated by centrifugation at 5000 x RCF, for 15 minutes at room temperature. The aqueous phases were transferred to a clean centrifuge tube. The extraction was then repeated two more times with equal volume of
10 phenol-chloroform, isoamyl alcohol (25:2421) and equal volume of chloroform. After the third extraction, the aqueous phases were mixed with 0.1 volume of 3M NaAc (pH 7.0) and 2 volumes of ethanol (95%), centrifuged for 20 minutes at 10,000 RCF and the resulting DNA
15 pellet was washed with 75% ethanol and then redissolved with a Tris-EDTA buffer consisting of 10mM Tris.Cl pH8.0, and 1mM EDTA pH8.0.

The resulting high molecular weight genomic DNA was then digested with DNase I to generate DNA
20 fragments sized approximately smaller than 5 Kb. The DNA fragments were extracted with phenol-chloroform and ethanol precipitation. The DNA pellet was dissolved in TE buffers (pH 8.0). The absorbance of the DNA at 260 nm and 280 nm was measured and the ratio of A_{260} to
25 A_{280} was 1.8. The DNA concentration was calculated based on OD_{260} ($1 OD_{260} = 50 \text{ ug/ml DNA}$).

B. Preparation of Genomic DNA-Method II

One gram of fresh porcine liver that had been cut
30 into small pieces was mixed with 20 ml of a lysate solution consisting of 5M Guanidine thiocyanate and 0.01M EDTA, pH 8.0. The resulting mixture was then homogenized in a blender. Glass particles of between 5um and 25um in diameter were added to the homogenate
35 to a final concentration of 5% in a glass particle-DNA

binding buffer consisting of 6M sodium perchlorate, 50 mM Tris.Cl, pH 8.0, and 10mM Trans-1, 2-diaminocyclohexane-N,N,N',N'-tetraacetic acid (TDTA). The resulting mixture was agitated at room temperature for 20 minutes and then centrifuged at 5000 x RCF for 1 min. The supernatant was then removed. The DNA and glass particle pellet was washed once with another 20 ml of glass particle-DNA binding buffer and once with 70% ethanol. The bound DNA was then eluted with 10 ml elution buffer consisting of 0.2M sodium perchlorate and 50mM Tris.Cl, pH 8.0, and 10 mM EDTA by agitation at room temperature for 20 minutes, and then spinning down the glass particles at 5000 x RCF, 1 minute. The supernatant was transferred to a clean tube and precipitated with ethyl alcohol. DNA was digested with DNase I to generate fragments less than about 5Kb. After extraction and precipitation, the DNA pellet was dissolved in TE buffer, pH 8.0. The ratio of A_{260} to A_{280} was 1.8. Using O.D. at 260 nm, the DNA concentration was calculated.

C. Preparation of RNA

Ten volumes of homogenization solution consisting of 126 g of LiCl and 360 g of Urea dissolved in 1 L of H_2O was added to one gram of finely chopped liver tissue to homogenize the tissue. The mixture was then homogenized for 2 min. at 20,000 RPM. The homogenate was then incubated at 4°C for 4 hours followed by centrifugation at 10,000 x g at 4°C for one hour. The precipitate was then dissolved in 2.5 ml of a resuspension buffer consisting of 10 mM Tris.Cl, pH 7.6; 1mM EDTA pH 8.0; and 0.5% SDS. An equal volume of a 25:24:1 solution of phenol, chloroform and isoamyl alcohol was then added to the resuspension. The mixture was then mixed at room temperature for 20 min.

14

followed by centrifugation at 3,000 x g at room temperature for 5 min. The aqueous upper phase was then transferred to separate container and a 1/10 volume of 3M NaAc(pH5.0) was added. Precipitation was then achieved by the addition of 2 volumes of ethyl alcohol followed by a 10-15 min. incubation at -20°C. The sample was then centrifuged for 20 min. at 5,000 x g at 5°C. The resulting RNA pellet was then redissolved in double distilled water. The concentration of RNA was then determined from the optical absorbance at 260 nm (1 OD₂₆₀ = 45 ug/ml RNA). The ratio of A₂₆₀ to A₂₈₀ was 1.95.

D. Preparation of Plasmid DNA

A plasmid-bearing bacteria strain, HB101 (Promega) transformed with the plasmid pUC 19 (GIBCO) was cultured in 500 ml LB containing 50 µg/ml Ampicillin till saturated. The culture was then centrifuged at 4000x g at 4°C for 15 min. The bacteria pellet was resuspended in 10 ml of STET (0.1 M NaCl; 10 mM Tris.Cl, pH8.0; 1 mM EDTA, pH8.0; 5% Triton X-100) and lysed by addition of 1 ml of 10mg/ml freshly made lysozyme. The bacteria lysate was boiled for 40 seconds and then set on ice. The lysate was then centrifuged at 12,000x g for 15 min. The supernatant was transferred and 2.5 M NaAc pH 5.2 was added to a final concentration of 0.25 M. Isopropanol was added to the supernatant (0.6 X volume), mixed, set at room temperature for five minutes and then centrifuged at 12,000x g for five minutes. The supernatant was removed and the plasmid DNA pellet was washed, using 75% alcohol. After removing the alcohol, the tube was set at room temperature for ten minutes to evaporate residual alcohol. The plasmid DNA was then redissolved in 1 ml TE containing 20µg/ml RNase and stored at 4°C.

Sources for nucleic acid other than the above mentioned methods include bacteriophage DNA, plasmid DNA, synthetic oligonucleotides, and nucleotide precursors as well as commercial suppliers, which are readily available to one of skill in the art. The prepared DNA or RNA can be digested with enzymes, e.g., DNase or RNase prior to its use in the barriers of the invention.

EXAMPLE 2

10

Representative Procedures for Preparing Nucleic Acid-Containing Compounds

A. DNA-Containing Lotion

15

Formula:	DNA	0.1g
	Polyethylene glycol 400	3.0g
	Citric acid	0.3g
	Water	79.6ml
	Ethanol (95%)	17.0ml

20

A DNA-containing lotion was prepared by slowly stirring 100 mg DNA (purified as described for Example 1A, Method I) into a solution containing 3.0 g of polyethylene glycol 400, 0.3 g of citric acid and 79.6 ml water. After the solution was mixed, 95% ethanol (17.0 ml) was added to the mixture followed by stirring at room temperature. The mixture optionally can be irradiated under ultraviolet lamp (30 watt, 254 nm) for 10 minutes, 50 cm from the light source. Such irradiation appears to improve the protective capability of the DNA lotion. The final concentration of polyethylene glycol 400 was 3.0% (W/V) and of citric acid was 0.3% (W/V).

35

B. DNA-Containing Cream

DNA (0.1 g) in 3 ml TE solution, purified as described for Example 1A, Method I, was added to Formula B (see below) with stirring. Formula A (see below) was then mixed with the DNA-formula B mixture and stirred at 75°C. The mixture was then cooled to

40

16

50°C. The formulations for formulas A and B were as follows:

Formula A

5	Stearic acid	3.0 g
	Laureate-23 (Brij 35 ICI)	0.35 g
	Ozaberite NO4	1.5 g
	Paraffin	1.5 g
	Mineral oil, visc 70	15.0 g
10	Petrolatum	4.0 g
	Cetyl alcohol	3.0 g

Formula B

15	GMS (glycerol monostearate)	3.0 g
	TEA (a Triethanolamine)	1.0 g
	Propylene glycol	3.0 g
	Methyl paraben	0.1 g
	Water	60.0 g
20	Perfume	0.5 g

Formula C DNA or RNA in 3 ml TE buffer 0.1 g

25

EXAMPLE 3

Representative Procedures for Preparing
Nucleic Acid-coated Surfaces

30

A. DNA-Coated Plastic and Polymers

DNA can be covalently cross-linked to free hydroxyl groups in plastic resins and polymers such as poly(vinyl alcohol) and cellulose.

35

To activate a hydroxyl-containing resin, 100 g of resin is added to a solution, at a temperature of about 15-20°C, consisting of 50 g of cyanogen bromide (CNBr) which had been previously dissolved in 100 ml of dimethyl-formamide. The pH of the resulting resin and CNBr mixture is immediately adjusted to between 10.5 and 11.5 with 2N NaOH. The mixture is then incubated at 15-20°C for 15 min. The treated resin is then

40

rinsed with cold distilled water for 5 min and then rinsed in 100 ml of 10mM KPO_4 , pH8.0, for 3 min in order to terminate the reaction.

DNA purified as described for Example 1A is preferably digested with DNase I sufficient to generate DNA fragments smaller than about 5 kilobases. The DNA fragments are then precipitated with ethanol. The DNA pellet is subsequently dissolved in 0.1 M potassium phosphate, pH 7.5, to a final concentration of 1% by weight.

For every 100 ml DNA solution from the above step, 100g of the treated plastic resin is added under constant agitation at a temperature of 4-10°C. The mixture is then incubated under constant agitation for 1 hour at 4-10°C. This procedure results in 95% of the CNBr activated hydroxyl groups in the plastic resin being covalently linked to DNA molecules. The reaction may be incubated longer at 4°C, e.g., up to 18 hours, to achieve higher coupling efficiencies.

The DNA coated plastic resin is separated from the remaining DNA solution by centrifugation (filtration could also be used), and then rinsed with 100 ml 0.1N NaOH for 5 min. Subsequently the DNA coated resin is further rinsed with 1000ml 10mM KPO_4 , at pH8.0 for 10 min. Finally, the coated resin is rinsed with distilled water for 10 min.

B. DNA-Coated Glass

A DNA containing genetic hazard ultraviolet radiation absorbing barrier was applied to a glass surface through the following steps:

1. A piece of clean glass such as vitreous silica (SiO_2) was coated, by spinning, spraying or dipping, with an aqueous dispersion containing about 0.01 to 0.1% by weight of DNA, purified as described in Example 1.
2. The DNA coated glass was then dipped into a second aqueous solution containing about 0.05 to 10% by weight of an organofunctional silane (gamma-amino

propyltriethoxy silane, gamma-glycidoxypropyltrimethoxy silane or N-vinyl-benzyl amino ethyl-3-trimethoxy silyl propylamine are all suitable) in a base or an acid catalyzed system at room temperature for a short period of time.

3. The coated glass was then dried at 100°C.

4. The coated glass was then heated in an air or oxygen atmosphere at the rate of about 10-20°C/min to 150°C.

10 The preferred amount of DNA in the coating is greater than about 50 μg per cm^2 of the glass. This desired amount of DNA in the coating is achieved by repeating steps 1 and 2 for a number of cycles prior to drying. The amount of DNA on the glass was monitored by
15 optical absorption at 260 nm. When the absorption at 260 nm by the coating was approximately 1 Unit/ cm^2 , the NA coating contained approximately 50 $\mu\text{g}/\text{cm}^3$.

EXAMPLE 4

20

Representative Procedure for Preparing Nucleic Acid-Coated Fibers

25 Fiber materials to be coated with NA in this procedure include natural fibers such as cotton thread and cellulose, as well as synthetic fibers, such as nylon, and glass fibers. The procedure described below

has been successfully performed with cotton thread, cellulose, nylon and glass fibers.

Each gram of dry fiber was dipped in at least 30 ml of a 0.01% ~ 0.1% DNA or RNA solution (isolated according to the procedures described for Example 1) for 1 minute at a temperature of 10-60°C. The preferred molecular weight of DNA was between 500 bp and 5000 bp and the preferred temperature range was between 22°C to 45°C.

The DNA or RNA-coated fiber was then dried at room temperature (for glass fiber at 100°C), approximately 30 minutes. The dried DNA or RNA coated fibers were then irradiated under a 30 watt ultraviolet lamp (UV wavelength 254 nm) for 5 min., 20 cm from the light source.

EXAMPLE 5

Representative Procedure for Preparing Nucleic Acid-Coated Matrices

A. Covalent Cross-linking of Double Stranded DNA Molecules to Free Hydroxyl Group-Containing Solid Matrices

Transparent polymer plastics, glass fiber, cellulose powder or cellulose fibers which contain free-hydroxyl groups have all been satisfactorily used in the following procedure to covalently link DNA to a matrix to create an NA-containing, genetic hazard UV radiation absorbing matrix. The procedure was as follows:

(1) The solid matrices were first incubated in 0.1 N NaOH for 5 min at a rate of 500 ml of 0.1 N NaOH for every 100 cm² of matrix material. The matrices were then rinsed with at least 1200 ml per 100 cm² of matrix of distilled water for at least 7 min., until

the pH of the treated matrices became neutral. The matrix material was then dehydrated in methanol.

(2) DNA purified as described for Example 1A, Method I was solubilized in TE, pH8.0, to a concentration of 0.1% (W/V). One volume of this DNA solution was then mixed with 4 volumes of a solution consisting of 50% by weight 1-cyclohexy-3-(2-morpholinoethyl) carbodiimide metho-p-toluene sulfonate (CMC) in 0.2 M sodium 2-(N-morpholino)ethanesulfonate, at pH 6.0.

(3) The treated solid matrices were submerged in the above DNA-containing mixture for 16 h at 20-24°C. The DNA coated matrices were then rinsed 3 times with distilled water at 5 min. intervals, and then allowed to dry.

B. Cross-Linking Single-Stranded NA to Plastic Matrices by Ultraviolet Radiation

Single-stranded DNA can be cross-linked to plastic matrices by ultraviolet radiation in the following procedure:

(1) Double-stranded DNA (prepared as described for method I, Example 1A) is first denatured into single-stranded DNA by either incubating the DNA in 0.1 M NaOH for 5 min. and adding HCl to final concentration 0.1 N and placing at room temperature 5 min. or by incubating the DNA at 98°C for 5 min., followed by immediately dipping the solution into ice water. The solution of single-stranded DNA is then brought to a concentration of 0.1% DNA by weight.

Single-stranded DNA may also be directly obtained from single-strand DNA bacteriophages such as bacteriophage F1.

(2) Plastic matrices, prepared as described above in Example 5A, are submerged into 95% ethanol at a

concentration of 500 ml of ethanol per 100g of plastic resin. After a 5 min. incubation at room temperature, the resin is then rinsed with 1 N HCl at 500 ml HCl per 100g of resin. The resin is then rinsed with distilled
5 water for 7 min. and then dried at 37°C.

(3) The treated resin is then submerged into the 0.1% DNA solution for 5 min. and then allowed to dry at room temperature.

(4) To cross-link the DNA to the resin, the dried
10 DNA-coated resin is irradiated for 15 min. under a 30 W Sylvania germicidal lamp, at a wave length range of 320-370nm. The lamp is placed about 22 cm from the treated resins.

(5) After the cross-linking, the DNA-coated
15 resins are rinsed with distilled water for 5 min. The DNA-coated plastic resins are then dried and stored at room temperature.

20 C. Cross-Linking RNA Molecules to Free Hydroxyl Group-Containing Matrices Catalyzed by Water Soluble Carbodiimide

RNA molecules can be cross-linked to hydroxy group-containing matrices by water soluble carbodiimide
25 in the following procedure:

(1) RNA molecules are purified from porcine liver tissue according to the procedure described for Example 1C, and stored in 70% ethanol at -80°C. The ethanol is removed by centrifugation immediately before using the
30 RNA molecules.

(2) Solid matrices of the type described above for Example 5 A are washed with methanol.

(3) A 0.1% by weight RNA solution is mixed with 0.2M sodium 2-(N-morpholino) ethanesulfonate, pH6.0, at
35 a ratio of 6:1. Carbodiimide is then added to final concentration of 7.2% W/W.

(4) 5000 cm² of solid matrices are submerged into 100 ml mixture described in step 3. The resulting mixture is then incubated for 24 hours at 22°C.

5 (5) Following the incubation, the RNA coated matrices are rinsed three times with distilled water, for 5 min. each rinse. The RNA coated matrices are then allowed to dry at room temperature and stored at room temperature.

10

EXAMPLE 6

Use of DNA-Containing Lotion to Reduce Sunlight-Induced Damage to DNA

15

A. DNA Restriction Endonuclease Digestion Test

One of the principal ways in which genetic hazard ultraviolet radiation causes cancer is through the formation of ultraviolet light-induced pyrimidine dimers in double stranded DNA. Once formed, thymidine and cytosine dimers can lead to permanent mutations in the genetic code when the DNA copied during cellular replication. This form of mutagenesis is widely believed to be the major route through which sunlight induces both melanoma and malignant non-melanoma skin cancers.

25

Thus, an assay that detects the formation of ultraviolet radiation-induced thymidine dimers in double stranded DNA is desirous in assaying the efficiency of a genetic hazard ultraviolet radiation filtration system. These experiments describe such an assay. The assay described in these experiments uses restriction endonucleases which recognize and cleave specific DNA sequences containing potentially dimerizing pairs of pyrimidine bases in double stranded DNA.

35

The basic concept of the assay is that the restriction enzymes will cut the DNA in a specifically recognized sequence if the bases are not dimerized, whereas pyrimidine dimers will prevent cleavage by the restriction endonuclease. To assay the extent of thymidine dimerization, one assays the efficiency of enzyme cleavage. These parameters are inversely proportional. To assay the efficiency of a genetic hazard ultraviolet radiation filter, one compares the enzymatic efficiency--and hence the amount of dimerization of the bases of control samples of double stranded DNA with test samples protected by the genetic-hazard UV filter. For example, samples could include: (1) double stranded DNA that is not exposed to ultraviolet radiation (no direct sunlight); (2) double stranded DNA exposed to genetic hazard ultraviolet radiation (direct sunlight), and (3) double stranded DNA exposed to genetic hazard UV radiation (direct sunlight) and filtered with a genetic hazard ultraviolet light absorbing filter.

The following experiment describes the use of the restriction endonuclease assay in the context of evaluating the efficiency of the DNA-containing, genetic hazard ultraviolet light-absorbing lotion prepared as described above in Example 2 A. The DNA-lotion was not irradiated.

Genomic DNA of human skin fibroblast cells was purified by conventional methods (Laird, P., Nucleic Acids Research, 19:4293, 1991) and sealed in transparent polyvinyl chloride (PVC) bags at a concentration of 50ug/ml in a buffer containing 10 mM Tris Cl pH 8.0. 20 mM EDTA pH 8.0. Eight such DNA containing plastic bags were used in the following experiment:

(1) Bag 1 received no exposure to sunlight.

(2) Bag 2 was exposed to sunlight with an uncoated PVC sheet interposed between the bag and the sun, from 9:00AM to 5:00PM, for 31 consecutive days during the month of July at an altitude of 40N.

5 (3) Bags 3-8 were exposed to sunlight under the identical conditions of bag 2, however for each of bags 3-8 a barrier PVC sheet was interposed between the bag and the sun; the barrier PVC sheet had been evenly coated with a given amount of a DNA-containing lotion
10 prepared as described above in Example 2A.

The thickness of the PVC used to form the sealed bags and the PCV barrier sheets was approximately 0.1-0.2 mm. The PCV barrier sheets were placed atop the DNA-containing PCV bags. Table 1 shows the amount of
15 DNA in ug per cm² of PVC sheet that was present on each barrier for bags 2-8.

(4) After exposure to the sunlight, all of the bags were treated via the following procedure:

(a) The DNA solution from each bag was
20 precipitated with ethanol. The DNA pellet was subsequently dissolved in TE buffer (10 mM Tris Cl pH 8.0, 1 mM EDTA pH 8.0). One microgram of DNA was digested with 10 units of the restriction enzymes Mse I
Hpa II at 37°C for 2 hours.

25 (b) Each digest was then separated on a 0.9% agarose gel (10 cm x 14 cm) run overnight (12-16 hours) at 0.5 volts/cm. Following the separation, the gels were stained 15 min with 0.5ug/ml ethidium bromide, and rinsed twice with 10 mM MgSO₄ for 15 min. each time.

30 (c) The efficiency of the DNA restriction digestion was determined as follows:

(i) The gels were photographed using reverse film under ultraviolet light to produce a black and white photograph of the bands in the gel.

(ii) The intensity (darkness due to the presence of DNA species) of each lane of the gel was then determined by a densitometer. The multiple and various sized species of DNA in the gel resulted in a DNA smear along the lane. In the presence of endonucleases, larger pieces of DNA are cut into smaller pieces, resulting in a shift in DNA-intensity of the gel from the upper half of the gel (high molecular weight species) to the lower half of the gel (low molecular weight species). If endonuclease activity is inhibited by UV-induced dimerization, this shift in intensity is reduced or lost.

(iii) The total intensity of each lane was defined as L_i , where i stands for each of bags 2-8. The intensity of the lower half of the lane (i.e., low molecular weight DNA) in each lane was defined as $L_{1/2,i}$. N was defined as the percentage intensity of the lower half lane for each bag (e.g., percentage of darkness in the lane due to low molecular weight species of DNA). A decrease in the N value of each bag (decreased intensity of lower molecular weight bands) relative to the control correlates to an amount of pyrimidine dimers formed in the DNA, and is mathematically calculated as $N = L_{1/2,i} / L_i$. Compared to the no-exposure control, the N value is smaller in those DNA samples which have had some damage (dimerization) and thus cannot be fully cut by the endonuclease.

Using this method of analysis, the control, non-UV exposed DNA was calculated as $N_0 = L_{1/2,0} / L_0$. The protection efficiency for each experimental bag was calculated as N_i , where $i=2,3,\dots,8$. The efficiency of each bag was then converted to a percentage of the control group via the formula: $N_i/N_0 \times 100\%$. The higher the N percentage, the greater the conservation of restriction sites and the less damaged the DNA, and

the greater the amount of protection afforded by the composition against UV-induced genetic hazard damage. The results are summarized in Table 1 below, and indicate a high protection efficiency (e.g., approximately 90% or greater) when 50 ug/cm² or greater amounts of DNA are used as a barrier.

TABLE 1

Conservation of Restriction Sites in DNA
Protected by DNA-containing Lotion

BAG NO.	Amount of DNA on PVC sheet (ug/cm ²)	Conservation of MSE I sites (N _i /N _o) x 100%	Conservation of Hpa II sites (N _i /N _o) x 100%
2	0	3.0	2.1
3	5	14.9	11.3
4	15	27.3	25.1
5	25	67.2	59.7
6	50	91.1	89.2
7	100	98.4	97.0
8	150	99.6	99.1

B. Plasmid DNA Transformation Efficiency Test

The extent of DNA damage caused by genetic hazard ultraviolet radiation can also be measured by assaying the transformation efficiency of plasmid DNA. When plasmid DNA is exposed to sunlight, the ultraviolet light-induced structure and function damage to the DNA results in a decrease in the efficiency with which the DNA can transform a bacterial cell.

Thus, one can assay both the extent of ultraviolet light-induced damage to DNA and the efficiency of protection afforded by ultraviolet light filters, by comparing the transformation efficiency of plasmid DNA exposed to direct sunlight, with that of DNA exposed to

filtered sunlight, and with that of DNA exposed to no sunlight.

In the following experiment, a 50ug/ml solution of plasmid pBR322 DNA was placed in 8 different PVC bags and used in the following experiment:

(1) Control bag 1 received no exposure to sunlight.

(2) Control bag 2 was exposed to sunlight with a non-coated PVC sheet interposed between the bag and the sun, from 9:00AM to 5:00PM, for 31 consecutive days during the month of July at an altitude of 40N.

(3) Bags 3-8 were exposed to sunlight under the identical conditions of bag 2, however for each of bags 3-8 a barrier PVC sheet was interposed between the bag and the sun. The barrier PVC sheet had been evenly coated with a given amount of the DNA-containing lotion preparation described above in Example 2A. The thickness and placement of the PCV sheets was as described above for Example 6A. Table 2 shows the amount of DNA in ug per cm² of PVC that was present on the barrier for bags 2-8.

(4) After exposure to the sunlight, all of the bags were treated via the following procedure:

(a) 10 ng of plasmid DNA from each bag was used to transform competent HB101 *E. coli* bacterial cells via the conventional CaCl₂ method (Ishiura et.al., Mol. Cell. Biol. 2::607-616, 1982).

(b) Following transformation, the bacteria were spread on LB agar plates containing 100ug/ml ampicillin which were then incubated overnight at 37°C.

(c) Resulting colonies were counted and their number converted to colonies per ug plasmid DNA. The results of the experiment are shown in the Table 2 which compares the transformation efficiency of bags 2-8 with that of the non-sunlight exposed control. An

amount of 50 ug/cm² or greater resulted in approximately 90% or greater colony formation, e.g. protection from UV-induced damage.

TABLE 2

5 Reduction in transformation efficiency related
to the amount of DNA-containing lotion used
to filter sunlight:

Transformation Assay

10

BAG NO.	Amount of DNA on PVC sheet (ug/cm ²)	Colony formation (percentage of non-exposed control)
2	0	5.1
3	5	15.3
4	15	30.0
5	25	71.4
6	50	89.7
7	100	95.9
8	150	98.5

15

20

EXAMPLE 7

Use of DNA-Coated Plastic Sheets to
Reduce Sunlight-Induced Damage to DNA

25

Table 3 summarizes the results of ultraviolet protection afforded by DNA-coated plastic sheets prepared as described above for Example 3B. The experiment was performed as described above for Example 6A with the amounts of DNA shown in Table 3, column 1 being the amount of DNA upon the coated plastic sheets.

30

TABLE 3

Conservation of Restriction Site in DNA
Protected by DNA Containing Lotion

Amount of DNA on plastic sheet (ug/cm ²)	Conservation of MSE I Site (N _i :N _o x100)	Conservation of Hpa II Sites (N _i :N _o x100)
0	9.5	7.0
15	29.4	31.2
50	85.2	89.1
150	98.6	99.6

Table 4 summarizes the results of ultraviolet protection afforded by DNA-coated plastic sheets prepared as described above for Example 3B. The experiment was performed as described above for Example 6B with the amounts of DNA shown in Table 4, column 1 being the amount of DNA upon the coated plastic sheets.

TABLE 4

Reduction in transformation efficiency related to the amount of DNA on DNA-coated plastic sheets used to filter sunlight:
Transformation assay

Amount of DNA on plastic sheet (ug/cm ²)	Colony formation (percentage of control)
0	4.3
15	35.8
50	85.9
150	95.6

EXAMPLE 8

Use of a DNA-Coated Glass Shield
to Reduce Sunlight-Induced Damage to DNA

Genomic DNA prepared as described for Example 6 was placed in quartz tubes and exposed to sunlight in the manner described for Example 6. DNA was coated onto clean glass as described for Example 3B, and the NA-

coated glass barriers were interposed between the quartz tubes and the sun as described for Example 6.

Protection of the DNA in the tubes was measured by the restriction enzyme digestion assay and transformation efficiency assay described for Example 3. The results are shown below in Tables 5 and 6, and demonstrate a high efficiency when about 50 ug/cm² or greater amounts of DNA are used as a barrier.

TABLE 5

Reduction in thymidine dimers related to the amount of DNA in DNA-coated glass used to filter sunlight:
Restriction endonuclease assay

Amount of DNA coated on transparent glass (ug/cm ²)	Thymidine dimers (Ni: N ₀ x100) MseI	Cytosine dimers (Ni: N ₀ x100) ApaI
0	3.2	2.9
15	30.1	27.2
50	90.2	88.1
150	98.0	97.8

TABLE 6

Reduction in transformation efficiency related to the amount of DNA on DNA-coated glass used to filter sunlight:
Transformation assay

Amount of DNA on transparent glass (ug/cm ²)	Colony formation (percentage of control)
0	2.8
15	33.5
50	89.6
150	98.7

Because many varying and different embodiments may be made within the scope of the inventive concept herein taught, and because many modifications may be made in the embodiments herein detailed in accordance with the descriptive requirement of the law, it is to be

understood that the details herein are to be interpreted as illustrative and not in a limiting sense.

I CLAIM:

1. A method for protecting a target from genetic hazard ultraviolet radiation comprising:

interposing a genetic hazard ultraviolet radiation-filtering barrier between a source of ultraviolet radiation and a target, the barrier comprising an effective ultraviolet radiation-filtering amount of nucleic acid.

2. The method of claim 1, wherein the nucleic acid is selected from the group consisting of natural and synthetic deoxy ribonucleic acid, ribonucleic acid, nucleotides, nucleosides, bases or derivatives thereof.

3. The method of claim 1, wherein said source of ultraviolet radiation is sunlight.

4. The method of claim 1, wherein said source of ultraviolet radiation is artificial UV light.

5. The method of claim 1, wherein the amount of nucleic acid in or on the barrier ranges from about 15 $\mu\text{g}/\text{cm}^2$ to about 150 $\mu\text{g}/\text{cm}^2$.

6. The method of claim 1, wherein said barrier comprises a solid or semi-solid surface.

7. The method of claim 6, wherein said solid or semi-solid surface comprises glass or plastic.

8. The method of claim 1, wherein said interposing comprises applying to the surface of a living organism an effective genetic hazard ultraviolet radiation-absorbing amount of nucleic acid in a carrier suitable for topical application.

9. The method of claim 8, wherein said carrier is a cream or lotion.

10. The method of claim 8, wherein said surface is animal skin.

11. The method of claim 1, wherein said target is a living organism.

12. The method of claim 11, wherein said target is a human.

13. The method of claim 1, wherein said interposing comprises placing over the eyes of a subject a pair of glasses containing an effective ultraviolet radiation-filtering amount of nucleic acid.

14. A method for protecting the eyes of a subject from damaging effects of sunlight or ultraviolet radiation, the method comprising placing over the eyes of a subject a pair of glasses containing an effective sunlight or ultraviolet radiation-filtering amount of nucleic acid.

15. A composition for preventing exposure to genetic hazard ultraviolet radiation comprising:

nucleic acid in an amount effective to selectively filter nucleic acid-damaging ultraviolet radiation;

a carrier suitable for topical application to skin; and

wherein the nucleic acid is dispersed or dissolved within said carrier.

16. The composition of claim 15, wherein said nucleic acid is selected from the group consisting of natural and synthetic DNA, RNA and oligonucleotides.

17. The composition of claim 16, wherein said nucleic acid is DNA.

18. The composition of claim 15, wherein said nucleic acid is selected from the group consisting of natural and synthetic nucleotides, nucleosides, bases and derivatives thereof.

19. An article for filtering genetic hazard ultraviolet radiation comprising:

an inert substance; and
nucleic acid, wherein an effective genetic hazard ultraviolet radiation-absorbing amount of nucleic acid is present in or on the surface of the inert substance.

20. The article of claim 19, wherein the amount of nucleic acid in or on the inert substance ranges from about 15 $\mu\text{g}/\text{cm}^2$ to about 150 $\mu\text{g}/\text{cm}^2$.

21. The article of claim 19, wherein the nucleic acid is selected from the group consisting of natural and synthetic DNA, RNA, and oligonucleotides.

22. The article of claim 21, wherein the nucleic acid is deoxyribonucleic acid.

23. The article of claim 21, wherein the nucleic acid is ribonucleic acid.

24. The article of claim 19, wherein the nucleic acid is selected from the group consisting of natural and synthetic nucleotides, nucleosides, bases, and derivatives thereof.

25. The article of claim 19, wherein the inert substance is plastic or glass.


26. The article of claim 19, wherein the inert substance is composed of fibers.

27. The article of claim 26, wherein the inert substance is composed of woven fibers.

28. A pair of glasses useful for protecting the eyes of a wearer from damaging effects of sunlight or ultraviolet radiation, said pair of glasses comprising an effective sunlight or ultraviolet radiation-filtering amount of nucleic acid in or on said glasses.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US95/08572

A. CLASSIFICATION OF SUBJECT MATTER IPC(6) :A61K 7/42 US CL :424/59 According to International Patent Classification (IPC) or to both national classification and IPC				
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) U.S. : 424/59, 424/60, 424/47 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) APS, CAPLUS, BIOSIS, EMBASE, MEDLINE, USPAT, JAPIO search terms: RNA, DNA, UV radiation, protection, sunscreen				
C. DOCUMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.		
X --- Y	US, A, 3,892,844 (ERLEMAN ET AL) 01 July 1975. See entire document, especially page 2.	1, 5-7, 10-14, 18 ----- 1, 5-7, 10-14, 18		
X --- Y	SCIENCE AND CULTURE, Vol. 45. No. 11, GHOSH ET AL, (1979), "Modification of ultraviolet damage in Rhizobium lupini under the influence of DNA and hydrolyzates of DNA and RNA", pages 450-452. See entire document.	1-5, 7, 8, 13 ----- 1-5, 7, 8, 13		
Y	US, A, 3,937,809 (JACOBI ET AL) 10 February 1976. See entire document, especially column 8, line 58.	1-28		
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.				
<table border="0"> <tr> <td> * Special categories of cited documents: "A" document defining the general state of the art which is not considered to be part of particular relevance "E" earlier document published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed </td> <td> "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family </td> </tr> </table>			* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be part of particular relevance "E" earlier document published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be part of particular relevance "E" earlier document published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family			
Date of the actual completion of the international search 16 AUGUST 1995		Date of mailing of the international search report 25 SEP 1995		
Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231 Facsimile No. (703) 305-3230		Authorized officer  ANDREW K. MILNE Telephone No. (703) 308-0196		

THIS PAGE BLANK (USPTO)